# Site-directed mutagenesis in hemoglobin

# Effect of some mutations at protein interfaces

B. Vallone, P. Vecchini, V. Cavalli and M. Brunori

Department of Biochemical Sciences and CNR Centre for Molecular Biology, University of Rome 'La Sapienza'. P.le Aldo Moro, 5, 00185 Roma, Italy

#### Received 31 March 1993

The role of selected amino acid residues in the monomer-monomer contacts of Hb A has been studied by site-directed mutagenesis of the  $\alpha$  chain bearing substitutions in the subunit surfaces. Mutation  $\alpha 38$ Thr $\rightarrow$ Trp induced a stabilization of tetrameric Hb-CO with a decrease of the  $K_d$  for the equilibrium  $\alpha \beta_2 \Leftrightarrow 2\alpha\beta$ , but had no effect on ligand binding. Mutation  $\alpha 40$ Thr $\rightarrow$ Arg resulted in a complete loss of cooperativity in ligand binding. Mutation  $\alpha 103$ His $\rightarrow$ Val had no noticeable effect. We also studied the behaviour of isolated, mutated  $\alpha$  chains with respect to self association: compared to wt  $\alpha$  chains, mutant  $\alpha 38$ Thr $\rightarrow$ Trp showed stabilization of the dimeric state and (at high protein concentration) a detectable amount of tetramers Mutant  $\alpha 103$ His $\rightarrow$ Val showed only a minor stabilization of the  $\alpha_2$  dimer.

Hemoglobin mutant; Subunit contact; Protein engineering

#### 1. INTRODUCTION

The allosteric behaviour of Hb, whereby the quaternary  $T\rightarrow R$  conformational transition modulates the oxygen affinity of the heme iron, depends on its assembly into an  $\alpha_2\beta_2$  heteroteramer [1,2]. Oligomeric vertebrate hemoglobins evolved from an ancestral monomeric globin via gene duplication and selection of complementary surfaces to promote specific aggregation. Other proteins acquire their characteristic biological activity only when associated into oligomers stabilized by weak forces between complementary surfaces, where packing of side chains is as tight as the interior of a protein [3].

Theories for the molecular basis of cooperative effects and allosteric control in the binding of  $O_2$  to Hb have been tested in the past by experiments carried out with natural mutants. In the last few years site directed mutagenesis has been successfully applied to the same problem [4–6], with original contributions concerning the role of distal site residues in controlling ligand affinity of the heme, and the significance of critical amino acid residues in the evolution of hemoglobins.

In the assembly of a functional tetramer, the  $\alpha$  and  $\beta$  chains interact with each other *via* two types of contacts called  $\alpha 1\beta 1$  and  $\alpha 1\beta 2$ , involving different amino acid side chains on each monomer [2,7,8]. Interface

Correspondence address: M. Brunori, Dipartimento di Scienze Biochimiche, Università di Roma 'La Sapienza', P.le Aldo Moro, 5, 00185 Roma, Italy. Fax: (39) (6) 4440 062.

 $\alpha 1\beta 1$  is rather rigid, since it involves a larger molecular contact (~ 2,000 Ų) and is unchanged in the T→R quaternary transition. Our attention was focussed on the  $\alpha 1\beta 2$  interface (~ 1,200 Å<sup>2</sup> in R and ~ 1,500 Å<sup>2</sup> in T), where the transition from T to R is associated to large conformational changes; moreover it is known that dissociation of Hb into  $\alpha\beta$  dimers involves cleavage along this interface [9]. Here two positions of the  $\beta$  chains ( $\beta$ 37 and  $\beta$ 40) establish extensive contacts with the partner  $\alpha$ chains both in Hb A and Hb H (the  $\beta_4$  tetramer, see [10]); in the  $\alpha$  chains the corresponding topological positions are occupied by Thr. The side chains  $\beta$ 37Trp and  $\beta$ 40Arg play a key role in keeping the  $\alpha$  and  $\beta$  chains associated in the tetrameric state, since two natural mutants at these positions, Hb Hirose ( $\beta$ 37Trp $\rightarrow$ Ser) and Hb Austin ( $\beta$ 40Arg $\rightarrow$ Ser), display an increased dissociation into dimers [11,12]. Moreover it is known that Trp at position  $\beta$ 37 is an useful optical probe to follow the quaternary state of Hb A, and therefore the type of contact at the  $\alpha 1\beta 2$  interface; the ligand induced change of the narrow-banded difference spectrum around 290 nm, has been assigned mostly to a change in the environment of  $\beta$ 37Trp as a result of the quaternary transition [13,14].

Using information from structural data [2] and from natural mutants we have engineered four mutations in the  $\alpha$  chains of Hb A to introduce new amino acid side chains which may enhance the stability of Hb heterotetramer and, more generally, help understanding the rules governing surface complementarity and recognition. Site directed mutagenesis of amino acid residues at the surfaces which mediate  $\alpha\beta$  aggregation may in

addition provide a clue to understand why the  $\alpha$  and  $\beta$  chains, though homologous in sequence and structure, display substantial differences in their association behaviour, given that isolated  $\beta$  chains can form a tetramer, while  $\alpha$  chains cannot [15,16].

# 2. MATERIAL AND METHODS

The Hb mutants were produced with the technique described by Kunkel et al. [17] using the gene of the  $\alpha$  chains, synthetized and expressed by Hoffman et al. [18]. This expression system allows the production in *E. coli* of fully functional tetrameric Hb. We used three 21-mers with a 3 bases mismatch for the single mutant, and a 62-mer with two 3 bases mismatches for the double mutant. No additional mutation was found in all mutants by resequencing the entire coding sequence.

Mutant  $\alpha 103 His \rightarrow Val$  was denatured by exposure to the detergents used to disrupt the cells, and the other three mutants contained a significant fraction of met Hb Iron oxidation and denaturation were avoided by growing the cells in a medium containing traces of carbon monoxide (CO) and breaking the cells by sonication. The bacterial extract was then centrifuged and the supernatant was passed through a CM-Sepharose column equilibrated with 10 mM phosphate buffer at pH 6.0; elution with a gradient from 10 mM phosphate at pH 6.0 to 70 mM phosphate at pH 7.0 resulted in purification of Hb as verified by SDS-PAGE. All the buffers and colums used in the purification contained traces of CO.

Splitting the  $\alpha_3\beta_2$  Hb tetramer was obtained by reaction with pHMB [19], purified chains were regenerated by incubation with  $\beta$ -mercaptoethanol

Rapid mixing experiments were carried with a Gibson-Durrum stopped-flow apparatus; flash photolysis experiments with the apparatus described by Brunori and Giacometti [20]. The fraction of slowly reacting (SRF) and quickly reacting form (QRF) with CO were determined after flash photolysis at different hemoglobin concentrations [21].

# 3. RESULTS AND DISCUSSION

#### 3.1. Mutant hemoglobins

#### 3.1.1. Molecular modelling

At the  $\alpha 1\beta 2$  interface, residues  $\beta 37$ Trp and  $\beta 40$ Arg play a very significant role in the stabilization of the tetramer; the corresponding topological positions on the  $\alpha$  chains are occupied by two Thr. Computations indicate that (respectively) 21% and 19% of the total non-bonded interactions with the facing a monomer can be attributed to these two residues ( $\beta$ 37Trp and  $\beta$ 40Arg). As shown in Fig. 1, panel a,  $\beta$ 37Trp makes contacts with 4 residues on the partner a chain, and its role in the the stability of the  $\alpha 1\beta 2$  contact is suggested by its invariance in all vertebrate hemoglobins [22]. On the contrary Thr in the corresponding position of the  $\alpha$ chains, although in a similar environment, establishes fewer contacts (Fig. 1, panel b). Simulation of the mutations:  $\alpha 38$ Thr $\rightarrow$ Trp,  $\alpha 41$ Thr $\rightarrow$ Arg and of the double mutant  $\alpha 38$ Thr $\rightarrow$ Trp/ $\alpha 41$ Thr $\rightarrow$ Arg using the program BRUGEL [23], indicated that these substitutions could all be accommodated in the  $\alpha 1\beta 2$  interface of both oxy and deoxy Hb; moreover mutation  $\alpha$ 38Thr→Trp (Fig.

1, panel c) was associated to a substantial stabilization of the interface and to an increase in the buried surface area (of 69 Ų per interface). Thus a Trp in this position can mimick the role of the Trp in the homologous site of the  $\beta$  chains, due to the pseudo-symmetry of the interface and to sequence homology between  $\alpha$  and  $\beta$  chains.

Mutation  $\alpha 103$ His $\rightarrow$ Val (interface  $\alpha 1\beta 1$ ) was produced to test the prediction by Chothia and Lesk, based on molecular modelling (personal communication), that this substitution could facilitate aggregation of isolated  $\alpha$  chains (see below).

## 3.1.2. Ligand binding

Table I contains a list of the  $\alpha$  chain Hb mutants prepared and characterized in this work, as well as information on the presence of heme-heme interactions and on the tetramer-dimer dissociation constant.

The time course of combination of deoxy Hb with CO is shown in Fig. 2 for the four Hb mutants. Mutant  $\alpha$ 41Thr $\rightarrow$ Arg and  $\alpha$ 38Thr $\rightarrow$ Trp/ $\alpha$ 41Thr $\rightarrow$ Arg are characterized by a monofasic, fast combination with CO ( $k_{\rm on} = 2.5 \times 10^6 \,\mathrm{M}^{-1} \,\mathrm{s}^{-1}$ , like the isolated Hb chains, [1]), indicating that introduction of an Arg at position  $\alpha$ 41 abolishes cooperativity in ligand binding (either by destabilizing the T state or by stabilizing the R state). On the other hand Hb  $\alpha$ 38Thr→Trp and Hb alo3His→Val both display a slow combination with CO, with an autocatalytic time course similar (but not identical) to that characteristic of Hb A (see Fig. 2). The kinetic time courses were analyzed with a 4-step irreversible Adair scheme, which allows to obtain a reliable estimate of the CO combination rate constants for the T state  $(k_{\rm T})$  and for the R-state  $(k_{\rm R})$ ; the result is reported in Table I in terms of the ratio  $(k_R/k_T)$ , which is a quantitative estimate of the kinetic cooperativity for CO [24, 25].

The oxygen binding isotherm was determined for the mutant Hb  $\alpha$ 38Thr $\rightarrow$ Trp; the result shows that oxygen binding is cooperative ( $n_{1/2} = 2.2$  and  $p_{1/2}(O_2) = 3.4$  mmHg), in agreement with the conclusion drawn from the kinetic data.

# 3.1.3. Measurement of tetramer-dimer dissociation constant

The finding that Hb  $\alpha 38$ Thr $\rightarrow$ Trp and Hb  $\alpha 103$ His $\rightarrow$ Val are both cooperative allows to estimate the equilibrium dissociation constant ( $K_d$ ) for the dimerization reaction ( $\alpha,\beta_2 \Leftrightarrow 2\alpha\beta$ ) of the CO-derivative of these mutants, using the photolysis method. It is known that after photodissociation of the Hb-CO complex with a long flash ( $\tau > \mu$ s) CO rebinds to Hb following a bifasic kinetics [1]. The relative amplitude of the two phases is protein concentration dependent [21], and the quickly reacting (QRF) and slowly reacting (SRF) forms correspond respectively to dimeric and tetrameric Hb. As shown in Fig. 3, HbCO  $\alpha 38$ Thr $\rightarrow$ Trp clearly

displays a tendency to aggregate into tetramers greater than wt HbCO, while the mutant Hb  $\alpha 103$ His $\rightarrow$ Val is indistinguishable from wt HbCO. The absolute values of the dissociation constants are reported in Table I.

These results on the stability of the liganded tetramers upon modification of the two  $\alpha\beta$  contact surfaces (i.e.  $\alpha 1\beta 1$  and  $\alpha 1\beta 2$ ) are in complete agreement with the literature [9]. In fact mutation (Thr $\rightarrow$ Trp) at the  $\alpha 1\beta 2$ 

a

b

С

Fig. 1. Stereo pictures of  $\beta$ 37Trp,  $\alpha$ 38Thr and  $\alpha$ 38Trp with their respective neighbours on the facing monomer in the following structures: (a and b) Human hemoglobin wt, R state. (c) Model of human Hb ( $\alpha$ 38Thr $\rightarrow$ Trp), R state. Trp and Thr are drawn with a thick line and their neighbours on the facing monomer with a thin line. These models, like all computations employed in this work, have been performed with the program BRUGEL, running in VAX/VMS environment and with an Evans & Sutherland PS390 station [23].

contact, known to be cleaved upon dissociation into dimers, is shown to induce a stabilization of the tetramer consistent with an increase in the number of molecular contacts established (in the mutant) by  $\alpha 38$ Trp with the complementary residues on the partner  $\beta$  chain (see Fig. 1, panel c). Mutation  $\alpha 103$ His $\rightarrow$ Val at the  $\alpha 1\beta 1$  contact, on the other hand, has no effect on dimerization of Hb-CO.

# 3.1.4. UV difference spectra

Information on the reorganization of  $\alpha 1\beta 2$  interface can be obtained by UV spectroscopy, since a narrow band around 290 nm was shown to reflect a perturbation of the environment of aromatics, particularly  $\beta 37$ Trp [13,14].

On the basis of the molecular modelling illustrated above (see Fig. 1), we expected the ligand-linked  $T\rightarrow R$  quaternary transition to be associated to a perturbation of all Trp at  $\alpha 1\beta 2$  interfaces (a total of 4 per tetramer) in mutant Hb  $\alpha 38$ Thr $\rightarrow$ Trp. Determination of the derivative of the oxy and deoxy UV spectra of Hb  $\alpha 38$ Thr $\rightarrow$ Trp and comparison with the spectra obtained for wt Hb, indeed shows that the amplitude of the fine structure around 290 nm (one maximum at 289 nm and two minima at 285 and 292.5 nm) is approximately doubled in the mutant Hb (see Fig. 4). This is consistent with expectation, and shows that the additional Trp at the  $\alpha 1\beta 2$  interface is a second optical probe of the quaternary change, which may be exploited for investigations of the interface dynamics.

## 3.2. Mutant $\alpha$ chains

The isolated mutant  $\alpha$  chains were studied to obtain information on the interface involved in  $\alpha$  chains dimerization, and on amino acid substitutions which may enhance association of the  $\alpha$  chains, estimated by analytical ultracentrifugation. As shown in Fig. 5, sedimentation coefficients were fully consistent with the literature for horse myoglobin [1] (monomeric  $s_{20,w} = 1.8$ ), and for *Scapharca inequivalvis* Hb (dimeric  $s_{20,w} = 2.8$ ) [26]. Sedimentation experiments with wt  $\alpha$  chains confirm some tendency to dimerize; the estimated dissocia-

Table I

Synopsis of the properties determined for the hemoglobins containing various mutant  $\alpha$  chains. Column labeled 'interface' indicates the surface region in which the mutation is situated. Ratio  $(k_R/k_T)$  determined from the time course of combination with CO (see text). Tetramer - dimer dissociation equilibrium constants  $(K_d)$ , expressed on heme basis.

Mutant	Interface	$k_{\rm R}/k_{\rm T}$	$K_{\rm d}$ for $\alpha_2\beta_2 \Leftrightarrow 2\alpha\beta$
$\alpha$ 38Thr $\rightarrow$ Trp	α1 <i>β</i> 2	30	0.65 μΜ
$\alpha$ 41Thr $\rightarrow$ Arg	$\alpha 1 \beta 2$	1	n.d.
$\alpha 103 \text{His} \rightarrow \text{Val}$	$\alpha  \beta $	44	$2.1 \mu M$
$\alpha$ 38Thr $\rightarrow$ Trp/ $\alpha$ 41Thr $\rightarrow$ Arg	$\alpha 1 \beta 2$	1	n.d.
wt Hemoglobin	-	47	$2.1 \mu M$

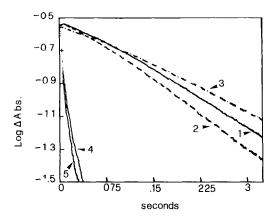


Fig. 2 Time course of the combination with CO of wt and mutant deoxy hemoglobins. Conditions: 0.1 M Bistris-Cl, pH 7.0, 20°C, CO=25  $\mu$ M after mixing, Hb 2  $\mu$ M after mixing. Trace 1, wt Hb A; trace 2, Hb  $\alpha$ 38Thr $\rightarrow$ Trp; trace 3, Hb  $\alpha$ 103His $\rightarrow$ Val; trace 4, Hb  $\alpha$ Thr41 $\rightarrow$ Arg; trace 5, Hb  $\alpha$ 38Thr $\rightarrow$ Trp/ $\alpha$ 41Thr $\rightarrow$ Arg. The dashed part of the trace in 4 and 5 represents the part of the reaction lost in the dead time of the instrument, given the fast combination with CO of the these two IIb mutants.

tion constant is however 5 times larger than the value reported by Valdes and Ackers [16]. More interesting is the finding that mutant  $\alpha 38 \text{Thr} \rightarrow \text{Trp}$  has a considerably greater tendency to dimerize, with a  $K_d$  at least 10 times smaller than that of wt  $\alpha$  chains; moreover at the highest concentrations (i.e. 13.5 mg/ml) we observed a sedimentation coefficient for this mutant significantly higher than the value expected for a dimer, suggesting a fraction of  $\alpha_4$  homotetramers.

On the other hand, the double mutant  $\alpha 38 \text{Thr} \rightarrow \text{Trp}/\alpha 41 \text{Thr} \rightarrow \text{Arg}$  and the single mutant  $\alpha 103 \text{His} \rightarrow \text{Val}$  (Fig. 5), are only slightly more associated than the wt  $\alpha$  chains; the single mutant  $\alpha 41 \text{Thr} \rightarrow \text{Arg}$  is like wt  $\alpha$  chains. Therefore, the mutation  $\alpha 38 \text{Thr} \rightarrow \text{Trp}$  at the  $\alpha 1\beta 2$  interface is the one that has a clearcut positive

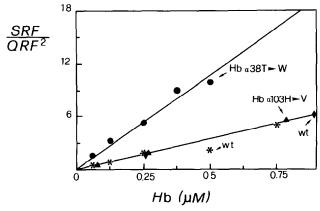


Fig. 3. Plot of the fraction SRF/QRF<sup>2</sup> as a function of tetrameric Hb A concentration; the slope of the straight line is equal to  $4/K_d$  for the dimerization reaction. Conditions: 0.1 M Bistris-Cl, pH 7.0, 20°C, CO =  $50\mu$ M. Symbols (mutants): •. Hb  $\alpha$ 38Thr $\rightarrow$ Trp; \*, wt Hb; •. Hb  $\alpha$ 103His $\rightarrow$ Val; •, wt Hb produced in E coli.

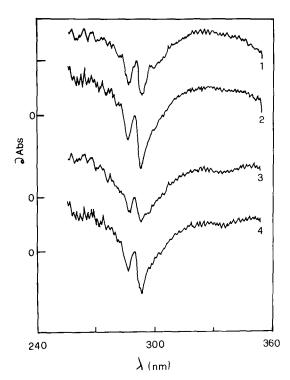


Fig. 4. UV region derivative spectra of oxy- and deoxyhemoglobins. Experimental conditions: 55  $\mu$ M Hb in 0.05 M Bistris-Cl, pH 7.4, t = 20°C. Derivative of the oxy and deoxy spectra were calculated using the first-derivative mode of the data manipulation program of the spectrophotometer (Olis Instruments modified Cary-14). 1, HbA-O2; 2, Hb ( $\alpha$ 38Thr $\rightarrow$ Trp)-O2; 3, deoxy Hb A; 4, deoxy Hb ( $\alpha$ 38Thr $\rightarrow$ Trp).

effect on association of the isolated  $\alpha$  chains, although some effect is also seen with the mutant  $\alpha 103$ His $\rightarrow$ Val.

These observations may suggest the following interpretation: wt  $\alpha$  chain dimers are formed via a subunit contact homologous to the  $\alpha 1\beta 1$  contact of Hb A, and mutation  $\alpha 103$ His $\rightarrow$ Val has a slight effect by eliminating an unfavourable contact at this interface. Introducing the mutation  $\alpha 38$ Thr $\rightarrow$ Trp, the contact  $\alpha 1\beta 2$  is also partially stabilized, and a fraction of  $\alpha_4$  tetramers is populated at high protein concentration. Arg in  $\alpha$ 41 as in the double mutant  $\alpha 38$ Thr $\rightarrow$ Trp/ $\alpha 41$ Thr $\rightarrow$ Arg, partially counterbalances the aggregating effect of Trp in α38, indicating that this interface is destabilized, consistently with the loss of cooperativity of the Hb carrying this mutation. Dimerization due to an  $\alpha 1\beta 1$ -like contact should not be affected by substitution  $\alpha$ 41Thr $\rightarrow$ Arg, and in fact the single mutant  $\alpha$  chains bearing an Arg in  $\alpha$ 41 shows the same behaviour as wt  $\alpha$  chains as far as dimerization is concerned.

This interpretation is in agreement with the modeling carried out by Vallone B., Schaad O. and Edelstein S.J. (unpublished data) who simulated an  $\alpha_4$  tetramer on the basis of the geometry of HbA, and found that the  $\alpha 1\beta 1$ -like contact involves more interacting surface than the interface topologically corresponding to  $\alpha 1\beta 2$ . There-

fore they formulated the hypothesis that  $\alpha$  chains dimerization was mediated by the contacts corresponding to interface  $\alpha 1\beta 1$  in Hb A; simulation of mutation  $\alpha 38 \text{Thr} \rightarrow \text{Trp}$  in the  $\alpha_4$  tetramer, seemed to stabilize the  $\alpha 1\beta 2$ -like contact.

#### 4. CONCLUSIONS

We have shown that it is possible to introduce a Trp at the  $\alpha 1\beta 2$  interface of human Hb at position  $\alpha 38$ ; this side chain is compatible with the overall thermodynamic stability of the T and R conformations, and does not alter the mechanism of molecular switch, since cooperativity in ligand binding is retained. The molecular contacts of Trp in  $\alpha 38$  seem analogous to those of  $\beta 37$ Trp, as predicted by simulation and confirmed by the observation that (similarly to  $\beta 37$ Trp) the optical spectrum of  $\alpha 38$ Trp is perturbed upon the T $\rightarrow$ R transition.

Mutation  $\alpha 38$ Thr $\rightarrow$ Trp is associated to stabilization of tetrameric HbCO, compared to wt Hb by 0.65 kcal/mol. Determination of the 3D structure of this mutant will allow a correlation of the measured stabilization free energy with the number of molecular contacts established by the indol side chain. It is very interesting that the isolated  $\alpha$  chains carrying this mutation are more tightly associated not only with the partner  $\beta$ 

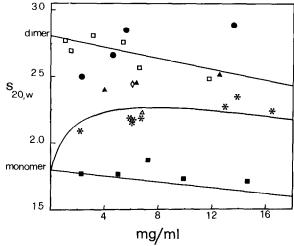


Fig. 5. Sedimentation coefficients  $s_{20,w}$  for hemoglobin  $\alpha^{SH}$  chains (wt and mutants) as the CO derivative, plotted as a function of the protein concentration. Sedimentation experiments were carried out on a Spinco Model E ultracentrifuge equipped with a RTIC unit at 52,000 rpm Conditions: Tris-Cl 0.1 M + 0.1 M NaCl, pH 7.4,  $t = 7-15^{\circ}$ C. The lines labeled 'monomer' and 'dimer' correspond to the values expected for dimeric and monomeric globins [15], corrected for the increased viscosity of the solution as the protein concentration increases, and are in agreement with values reported for horse Mb [1] and the dimeric hemoglobin from Scapharca inequivalvis [26]. The curve represents the expected  $s_{20,w}$  for a  $K_d = 0.49$  mM for the equilibrium  $\alpha_2 \Leftrightarrow 2\alpha$ . Symbols:  $\blacksquare$ , horse Mb;  $\square$ , Scapharca mequivalvis dimeric Hb; \*, wt Hb  $\alpha^{SH}$  chains; •,  $\alpha^{SH}$  chains  $\alpha_{38}$ Thr $\rightarrow$ Trp;  $\triangle$ ,  $\alpha^{SH}$  chains  $\alpha_{41}$ Thr $\rightarrow$ Arg;  $\triangle$ ,  $\alpha^{SH}$  chains  $\alpha_{103}$ His $\rightarrow$ Val; •,  $\alpha^{SH}$  chains  $\alpha_{38}$ Thr $\rightarrow$ Trp/ $\alpha_{41}$ Thr $\rightarrow$ Arg.

chains (across the  $\alpha 1\beta 2$  interface), but also with other  $\alpha$  chains, given that full  $\alpha_2$  dimers and a fraction of  $\alpha_4$  tetramers are detected by ultracentrifugation.

A dramatic effect on the properties of Hb was observed upon introduction of a second Arg at position C6 of interface  $\alpha 1\beta 2$ , as in mutant  $\alpha 41 \text{Thr} \rightarrow \text{Arg}$  which completely loses coperativity in ligand binding. This effect was not predicted by modelling, possibly due to underestimation of the electrostatic component of the nonbonded energy in the potential energy function used. Interestingly also Hb H (i.e. the  $\beta$  tetramer), that has two Arg at position C6, is noncooperative in ligand binding [1], suggesting that the T conformation is not allowed in the presence of two Arg in the  $\alpha 1\beta 2$  interface, possibly because  $\alpha 41 \text{Arg}$  (or  $\beta 40 \text{Arg}$ ) makes unfavourable contacts with  $\beta 145 \text{Tyr}$  across the  $\alpha 1\beta 2$  interface.

In summary we obtained novel information on the role of some amino acid side chains at the molecular contacts between hemoglobin subunits, and engineered improved protein-protein surfaces, which lead to enhanced association of one mutant  $\alpha$  chain ( $\alpha$ 38Thr $\rightarrow$ Trp) with another  $\alpha$  chain or with its partner  $\beta$  chain. This is yet another example that point mutations at a physiologically important interface promote association, leading to evolution of complementary surfaces mediated by a few mutational events, and permitting fine tuning of biological function.

Acknowledgements We wish to gratefully acknowledge K. Nagai and N.H. Komiyama (LMB-MRC, Cambridge, UK) for their invaluable help in the production of Hb mutants, initially carried out at LMB-MRC. We also thank C. Chothia and A.M. Lesk (ibidem) for their stimulating suggestions and A. Bellelli, R. Ippoliti, E. Lendaro and A. Brancaccio (Rome) for their help in some hemoglobin manipulations. We are very grateful to Prof. M. Tomasi and Mr. B. Volpi of ISS for their help in optimizing bacterial growth conditions. Permission to use the Hb expression system in E. coli for scientic purposes given by Somatogen Inc. is gratefully acknowledged. B.V. was supported by an EMBO short term fellowship and by the Società Italiana di Biochimica. Grants from the MURST (40% Liveprotein) and the CNR (Progetto Strategico 'Trasduzione') are gratefully ackowledged.

# REFERENCES

[1] Antonini, E and Brunori, M. (1971) Hemoglobin and Myoglobin in their reactions with ligands, North Holland. Amsterdam

- [2] Perutz, M.F. and Fermi, G. (1981) Hemoglobin and Myoglobin, Atlas of Protein Sequence and Structure Vol. 2, Oxford University Press. New York
- [3] Janin, C and Chothia, C. (1990) J. Biol. Chem. 256, 16027– 16030.
- [4] Nagai, K. Perutz, M.F. and Poyart, C (1985) Proc. Natl. Acad. Sci. USA 82, 7252–7255.
- [5] Nagai, K. Luisi, B. Shih, D.T.-H. Miyazaki, G., Imai, K., Poyart, C., De Young, Kwiatowsky, L., Noble, R.W., Lin S.-H. and Yu N.T. (1987) Nature 329, 858–860.
- [6] Imai, K. Shih, D.T.-B. Tame, J. Nagai, K. and Miyazaki G (1989) Protein Seq. Data Anal. 2, 81–86.
- [7] Baldwin, J and Chothia, C. (1979) J. Mol. Biol. 129, 175-220.
- [8] Lesk, A.M. Janin, J. Wodak, S. and Chothia C. (1985) J. Mol. Biol. 183, 267–270
- [9] Edelstein, S.J. Rehmar, M.J. Olson, J.S. and Gibson, Q. (1970) J. Biol. Chem. 245, 4372–4381.
- [10] Arnone, A. and Briley, P.D. (1987) in. Biochemical and Clinical Aspects of Hemoglobin Abnormalities (W S. Caughey, Ed.) pp. 93-107. Academic Press. New York
- [11] Sasaki, J. Imamura, T. Yanase, T. Atha, D.H. Riggs, A. Bona-ventura, J. and Bonaventura C. (1978) J. Biol. Chem. 253, 87–94.
- [12] Moo-Penn, W.F. Johnson, M.H. Bechtel, K.C. Jue, D.L. Therrell, B.L. Schmidt, R.M. (1977) Arch. Biochem. Biophys 179, 86-94.
- [13] Imai, K. and Yonetani, T. (1973) Biochem. Biophys. Res. Commun 50, 1055–1060.
- [14] Perutz, M F. Ladner, J.E. Simon, S R and Chien Ho (1974) Biochemistry 13, 2163–2173.
- [15] Bucci, E. Fronticelli, C. Chiancone, E. Wyman, J. Antonini, E. and Rossi-Fanelli, A. (1965) J. Mol. Biol. 12, 183–192.
- [16] Valdes, R. and Ackers, G. (1977) J. Biol Chem. 252, 74-81.
- [17] Kunkel T.A. Roberts and J.D. Zakour, R.A. (1987) Methods Enzymol. 154, 367-382.
- [18] Hoffman, S.J. Looker, D.L. Roerich, J.M. Cozart, P.E. Durfee, S.L. Tedesco, J.L. and Stetler G (1990) Proc. Natl. Acad. Sci. USA 87, 8521–8525
- [19] Bucci, E. and Fronticelli, C (1965) J. Biol. Chem. 240, 551-553.
- [20] Brunori, M and Giacometti, G.M. (1981) Methods Enzymol. 76, 582–595.
- [21] Gibson Q. H. and Antonini E. (1967) J. Biol. Chem. 242, 4678– 4681.
- [22] Dickerson, R. E. and Geis, 1 (1983) Hemoglobin, Structure, Function Evolution and Pathology, Benjamin/Cummings, USA.
- [23] Delhaise, P. Bardiaux, M. and Wodak, S.J. (1984) J. Mol. Graphics 2, 103–106
- [24] Szabo, A. (1978) Proc. Natl. Acad. Sci. USA 75, 2108-2111.
- [25] Hopfield, J.H. Shulman, R.G. and Ogawa, S. (1971) J. Mol. Biol. 61, 424–433.
- [26] Chiancone, E. Vecchini, P. Verzili, D. Ascoli, F. and Antonini, E. (1981) J. Mol. Biol. 152, 577–592.